Name / Position
Lin, Hong-Liang / Assistant Professor

## Office Address

School of Pharmacy, College of Pharmacy, Kaohsiung Medical University. 100, Shih-Chuan 1<sup>st</sup> Rd. Kaohsiung, Taiwan, 80708

- E-mail : <u>hlglin@kmu.edu.tw</u>
- Tel: 886-7-3121101 ext. 2565
- Fax : 886-7-3210683
- The Highest Education Degree

Ph.D. in School of Pharmacy, Taipei Medical University, Taipei, Taiwan.

- Teaching Courses
  - Pharmaceutical Physical Chemistry
  - Pharmaceutical Dosage Forms
  - Industrial Pharmacy
  - Drug Verification
  - Special Topics on Verification and Validation of Pharmaceutical Manufacturing
  - Development and Validation of Health Food
  - Special Topics on Solid Dosage Forms
  - Special Topics on Advanced Industrial Pharmacy

## • Research Area

- Development and design of drug dosage form
- Drug delivery system
- Pharmaceutics
- Industrial pharmacy
- Modified release dosage forms (extended-release, prolonged-release, controlled-release, slow release and sustained-release)

## Recent Publications

- Lin Hong-Liang, Ho Hsiu-O, Chen Chi-Chia, Yeh Ta-Shuong, Sheu Ming-Thau\*. (2008) Process and formulation characterizations of the thermal adhesion granulation (TAG) process for improving granular properties. *Int. J. Pharmaceut.* 357: 206–212. [SCI]
- Lin Hong-Liang, Lin Shyr-Yi, Lin Ying-Ku, Ho Hsiu-O, Lo Yo-Wen, Sheu Ming-Thau\*. (2008) Release characteristics and in vitro-in vivo correlation of pulsatile pattern for a pulsatile drug delivery system activated by membrane rupture via osmotic pressure and swelling. *Eur. J. Pharm. Biopharm.* 70: 289–301. [SCI]
- Lin Shan-Yang\*, Cheng Wen-Ting, Wei Yen-Shan, <u>Lin Hong-liang</u>. (2011) DSC-FTIR microspectroscopy used to investigate the heat-induced intramolecular cyclic anhydride formation between Eudragit E and PVA copolymer. *Polym. J.* 43: 577–580. [SCI]
- 4. Wu Tieh-Kang, Lin Shan-Yang\*, <u>Lin Hong-liang</u>, Huang Yu-Ting. (2011) Simultaneous DSC-FTIR microspectroscopy used to screen and detect the co-crystal formation in real time. *Bioorg*.



Med. Chem. Lett. 21: 3148-3151. [SCI]

- Lin Hong-liang, Lin Shan-Yang\*, Lin Chih-Cheng, Hsu Cheng-Hung, Wu Tieh-Kang, Huang Yu-Ting. (2012) Mechanical grinding effect on thermodynamics and inclusion efficiency of loratadine–cyclodextrin inclusion complex formation. *Carbohyd. Polym.* 87: 512-517. [SCI]
- Zhang Gang-Chun, <u>Lin Hong-Liang</u>\*, Lin Shan-Yang\*\*. (2012) Thermal analysis and FTIR spectral curve-fitting investigation of formation mechanism and stability of indomethacin-saccharin cocrystals via solid-state grinding process. *J. Pharmaceut. Biomed. Anal.* 66: 162-169. [SCI]
- Hsu Po-Chun, <u>Lin Hong-Liang</u>, Wang Shun-Li<sup>\*\*</sup>, Lin Shan-Yang<sup>\*</sup>. (2012) Solid-state thermal behavior and stability studies of theophylline-citric acid cocrystals prepared by neat cogrinding or thermal treatment. *J. Solid State Chem.* 192: 238-245. [SCI]
- 8. <u>Lin Hong-Liang</u>, Zhang Gang-Chun, Hsu Po-Chun, Lin Shan-Yang<sup>\*</sup>. (2013) A portable fiberoptic Raman analyzer for fast real-time screening and identifying cocrystal formation of drug-coformer via grinding process. *Microchem. J.* 110: 15-20. **[SCI]**
- Lin Hong-Liang, Hsu Po-Chun, Lin Shan-Yang\*. (2013) Theophylline-citric acid co-crystals easily induced by DSC-FTIR microspectroscopy or different storage conditions. *Asian Journal of Pharmaceutical Sciences.* 8: 19-27. [EI]
- Lee Lin-Wen, Hung Sheng-Feng, <u>Lin Hong-Liang</u>, Ho Hsiu-O, Sheu Ming-Thau\*. (2013) Development of Timely Controlled-Release Systems for Chronotherapy of Propranolol with Minimization of the pH Effect in the Simulated Gastrointestinal Medium. *J. Food Drug Anal.* 21: 115-125. [SCI]
- Lin Hong-Liang\*, Wu Tieh-Kang, Lin Shan-Yang\*\*. (2014) Screening and characterization of cocrystal formation of metaxalone with short-chain dicarboxylic acids induced by solventassistedgrinding approach. *Thermochim. Acta.* 575: 313-321. [SCI]
- Lin Hong-Liang\*, Zhang Gang-Chun, Huang Yu-Ting, Lin Shan-Yang\*\*. (2014) An Investigation of Indomethacin–Nicotinamide Cocrystal Formation Induced by Thermal Stress in the Solid or Liquid State. J. Pharm. Sci. 103: 2386-2395. [SCI]
- Lin Hong-Liang\*, Zhang Gang-Chun, Lin Shan-Yang\*\*. (2015) Real-time co-crystal screening and formation between indomethacin and saccharin via DSC analytical technique or DSC–FTIR microspectroscopy. J. Therm. Anal. Calorim. 120: 679-687. [SCI]
- Kao Chi-Yu, Huang Huai-Han, Huang Yu-Ting, <u>Lin Hong-Liang</u>, Lin Shan-Yang\*. (2015) Thermoanalytical and spectroscopic studies on amorphization and phase transition of amorphous indomethacin prepared by two melt-cooling processes. *Sci. Letter. J.*, 4: 148
- Lin Hong-Liang, Chi Ying-Ting, Huang Yu-Ting, Kao Chi-Yu and Lin Shan-Yang\*. (2015) DSC-FTIR Combined Approaches Used to Simultaneously Prepare/Determine the Amorphous Solid Dispersions of Indomethacin/Soluplus in Real-time. *EC Pharmaceutical Science* 2.1: 183-193.
- Lin Shan-Yang\*, <u>Lin Hong-Liang</u>, Chi Ying-Ting, Huang Yu-Ting, Kao Chi-Yu, Hsieh Wei-Hsien. (2015) Thermoanalytical and Fourier transform infrared spectral curve-fitting techniques used to investigate the amorphous indomethacin formation and its physical stability in Indomethacin-Soluplus<sup>®</sup> solid dispersions. *Int. J. Pharmaceut.* 496: 457-465. [SCI]

- Lin Shan-Yang\*, <u>Lin Hong-Liang</u>, Hung Ru-Ying, Huang Yu-Ting and Kao Chi-Yu. (2015) Effect of Povacoat or Soluplus on Solid-State Characterization of Indomethacin-Nicotinamide Co-Crystal Formation. *Pharm. Anal. Acta.* 6:8 402-408.
- Lin Shan-Yang\*, <u>Lin Hong-Liang</u>, Chi Ying-Ting, Hung Ru-Ying, Huang Yu-Ting, Kao Chi-Yu, Hsieh Wei-Hsien. (2016) Povacoat affecting solid-state polymorphic changes of indomethacin after co-evaporation from different types of solvents via conventional and microwave drying techniques. *Asian Journal of Pharmaceutical Sciences*. 11(3): 376-384.[EI]
- Wei-Hsien Hsieh\*, Wen-Ting Cheng, Ling-Chun Chen, <u>Hong-Liang Lin</u> and Shan-Yang Lin\*. (2017) Non-isothermal Dehydration Kinetics of Glucose Monohydrate, Maltose Monohydrate and Trehalose Dihydrate by Thermal Analysis and DSC-FTIR Study. *J. Biomed. Pharm. Sci.* 1:101.[In Press]